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### Advancements in Nanoengineering: An analytical Study on the Synthesis, Characterization, and Diverse Applications of TiO<sub>2</sub>-Graphene Nanocomposites

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#### ABSTRACT

Nanoengineering wields the power to redefine the boundaries of human innovation, enabling the precise manipulation of materials at the nanoscale. This capability not only fuels groundbreaking advancements in medicine, electronics, and energy but also holds transformative potential for enhancing human health and quality of life. This research paper delves into the dynamic realm of nanomaterials by exploring the synthesis, characterization, and multifaceted applications of TiO<sub>2</sub>-Graphene composite materials. The synergistic combination of titanium dioxide (TiO<sub>2</sub>) and graphene has garnered significant attention for its unique properties, offering a platform for enhanced functionality and performance. The synthesis process is meticulously investigated, aiming for precise control over the nanocomposite's composition and morphology. Comprehensive characterization techniques, including spectroscopy, microscopy, and thermal analysis, provide insights into the structural and chemical properties, enabling a thorough understanding of the material. The paper also delves into the various TiO<sub>2</sub>-Graphene nanocomposites have uses in many different fields. From energy storage and conversion devices to environmental remediation and catalysis, the nanocomposite's versatile nature is explored. The exceptional conductivity of graphene coupled with the photocatalytic properties of TiO<sub>2</sub> opens avenues for efficient solar energy utilization and pollutant degradation. Furthermore, the nanocomposites exhibit promising potential in biomedical applications, showcasing their versatility in addressing contemporary challenges. Elevate textile effluent remediation with our advanced TiO<sub>2</sub>-Graphene Nanocomposites. Harnessing Nanoengineering, we offer efficient, sustainable solutions for rapid dye degradation and environmental conservation.

**Keywords:** TiO<sub>2</sub>-Graphene Nanocomposites, Graphene Oxide, Photocatalysis

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## 1. INTRODUCTION:

Nanotechnology, a frontier field in materials science, has revolutionized the landscape of material design and application (Zhang & Li, 2021). As nanoengineering continues to push the limits of what is possible (Wang & Liu, 2014), it stands poised to revolutionize industries and contribute to a future where the positive impact on human well-being is both profound and far-reaching (Chen et al., 2018). Within this realm, the synthesis of nanocomposites has emerged as a fascinating avenue, offering unprecedented possibilities for tailoring materials with enhanced properties (Zhao et al., 2018). Among these, the integration of titanium dioxide (TiO<sub>2</sub>) with graphene has garnered considerable attention for its synergistic effects, combining the unique characteristics of both materials (Liu et al., 2020). TiO<sub>2</sub>, renowned for its photocatalytic process, and graphene, celebrated for its exceptional conductivity, together form a powerful duo that holds promise for diverse applications. This research contributes to the expanding knowledge base in nanomaterial science, providing a comprehensive overview of TiO<sub>2</sub>-Graphene nanocomposites and their potential impact on advancing technological and environmental solutions.

In the realm of nanomaterials, the creation of nanocomposites of TiO<sub>2</sub>-Graphene stands as a compelling intersection of innovation and scientific ingenuity (Zhang & Li, 2021). This unique amalgamation of titanium dioxide (TiO<sub>2</sub>) and graphene represents a promising class of nanomaterials with distinctive properties that transcend the individual strengths of their constituent components (Wang & Liu, 2014). The synthesis process holds a pivotal role in sculpting these nanocomposites, attempting to achieve a delicate equilibrium between regulated composition and tailored morphology (Chen et al., 2018). As researchers delve into the intricacies of crafting TiO<sub>2</sub>-Graphene nanocomposites, the goal is not merely to combine materials at the nanoscale but to harness a synergistic interplay that unlocks unprecedented potential for applications ranging from energy storage to environmental remediation (Zhao et al., 2018). This introductory exploration aims to unravel the nuanced art of synthesizing TiO<sub>2</sub>-Graphene nanocomposites, underscoring the importance of methodical design and characterization for unlocking their full capabilities in the realm of advanced materials science.

The chemical formulation and synthesis process of TiO<sub>2</sub>-Graphene nanocomposites involve meticulous control over the combination of titanium dioxide (TiO<sub>2</sub>) and graphene at the nanoscale to harness their synergistic properties. The formulation begins with the selection of appropriate precursor materials for TiO<sub>2</sub> and graphene, considering their compatibility and reactivity.

For TiO<sub>2</sub>, common precursors include titanium alkoxides or titanium salts, while graphene may be derived from graphene oxide or graphene nanoplatelets. The chosen synthesis route aims to achieve a uniform dispersion of both components and enhance their interaction.

Sol-gel is a technique that frequently employed approach where in titanium precursors are hydrolyzed and then polymerized, forming a gel-like structure. This gel can be combined with graphene through mechanical mixing or sonication, ensuring a homogenous distribution of graphene within the TiO<sub>2</sub> matrix. Subsequent heat treatment or calcination drives the removal of organic components and crystallization of TiO<sub>2</sub>, contributing to the formation of the nanocomposite structure.

Alternatively, graphene oxide can be reduced to graphene by chemical reduction techniques using reducing agents such as hydrazine or ascorbic acid, and then graphene is incorporated into the TiO<sub>2</sub> matrix. The final properties of the nanocomposite can be influenced by controlling the degree of reduction of graphene oxide by this approach.

Characterization techniques, such as transmission electron microscopy (TEM), Fourier-transform infrared spectroscopy (FTIR), and X-ray diffraction (XRD) are crucial in assessing the chemical structure, composition, and morphology of the synthesized TiO<sub>2</sub>-Graphene nanocomposites.

The chemical formulation and synthesis process play a pivotal role in tailoring the properties of TiO<sub>2</sub>-Graphene nanocomposites, influencing their applications in photocatalysis. As researchers refine and advance these synthesis techniques, the potential for harnessing the full capabilities of these nanocomposites in various technological domains continues to expand.

Through numerous mechanisms, photocatalysts benefit energy and the environment. Semi-conductor heterogeneous photocatalyst TiO<sub>2</sub> is popular for its photostability, electrical and surface qualities, non-toxicity, cost-effectiveness, and eco-friendliness (Kumar and Devi, 2011; Schneider et al., 2014). Low visible light photoconversion efficiency, quantum yield, and recombination rate render TiO<sub>2</sub> photocatalysis inappropriate for commercial usage (Guo et al., 2019b).

The 2D honeycomb lattice structure of graphene, a single-atom-thick carbon sheet, offers amazing physical and chemical capabilities. In the recent decade, TiO<sub>2</sub>-graphene nanocomposites have been created, described, and employed in photocatalysis. The nanocomposites' optical, electrical, and physicochemical properties are good (Williams et al., 2008; Liu 2011, Perera 2012, Sun 2013, Chen 2020). This work examines TiO<sub>2</sub>/graphene interface charge transfer kinetics, although graphene derivative-based photocatalysts have several uses. Good physicochemical and electrical qualities are present (Williams et al., 2008; Liu 2011, Perera 2012, Sun 2013, Chen 2020). Although photocatalysts based on graphene derivatives have various applications, this work investigates the kinetics of charge transfer at the TiO<sub>2</sub>/graphene interface. Many approaches improve TiO<sub>2</sub>-graphene hybrid interaction. Graphene nanosheet defect engineering, functionalization, and doping, and TiO<sub>2</sub> facets and morphological customization are examples.

TiO<sub>2</sub> photocatalysis research has yielded several uses since the 1960s. Kato and Mashio discovered liquid phase tetralin oxidation under UV light in 1964, starting TiO<sub>2</sub> photocatalysis. Yoneyama et al. (1972), Steinbach (1967), and Wolkenstein (1973) described the use of UV-illuminated semi-conducting TiO<sub>2</sub> in gas-phase photocatalytic reactions. In 1972, Honda and Fujishima pioneered the process of electrolyzing water into H<sub>2</sub> and O<sub>2</sub> using a Pt cathode and a TiO<sub>2</sub> photoanode exposed to UV light. In 1977, Schrauzer and Guth discovered that water in Pt or Rh-loaded TiO<sub>2</sub> is killed by light. Pt metal is reduced by TiO<sub>2</sub> electrons, although holes remain and oxidize at the surface (Schrauzer and Guth, 1977).

TiO<sub>2</sub> has a 3.0-3.2 eV bandgap. Plasmaez et al. (2012) and Ganguly (2017) demonstrated that UV light below 400 nm forms Valence Band holes by attracting

electrons to the Conduction Band. By recombining with Valence Band holes or photocatalysis, Conduction Band electrons may carry charge. Lower charge separation and migration duration diminish Recombination (Padmanabhan, 2020a, 2020b; Ganguly et al., 2017). An electron and a hole called an exciton oxidize and decrease close to the semiconductor surface. Hydroxyl radicals are produced by Valence band excitons and surface water molecules (Mathew et al., 2018). Direct hole-reactant oxidation. Superoxide anions are formed by oxygen and conduction band electrons (Panneri et al., 2016).

## **Characterization and applications**

Beyond the synthesis, a detailed characterization of the resulting nanocomposites is imperative to unravel their structural intricacies and elucidate the synergistic effects that contribute to their unique properties (Zhao et al., 2018; Belchi et al., 2020; Ruidíaz-Martínez et al., 2022). Spectroscopy, microscopy, and thermal analysis techniques are employed to scrutinize the structural, chemical, and thermal attributes, providing a holistic understanding of the synthesized TiO<sub>2</sub>-Graphene nanocomposites (Giovannetti et al., 2021; Sagadevan et al., 2023).

As we embark on this scientific journey, the potential applications of TiO<sub>2</sub>-Graphene nanocomposites beckon exploration. From energy storage and conversion devices to environmental remediation and catalysis, the versatility of these nanocomposites positions them as transformative agents across multiple fields (Farooq et al., 2022; Yousefzadeha et al., 2023). The exceptional electrical conductivity of graphene coupled with the photocatalytic prowess of TiO<sub>2</sub> opens avenues for efficient solar energy utilization in textile effluent remediation, while their catalytic properties hold promise for environmental sustainability (Arif Sher Shah et al., 2024; Padmanabhan et al., 2023).

By shedding light on the synthesis, characterization, and various applications of TiO<sub>2</sub>-Graphene nanocomposites, this research aims to contribute to the changing field of nanomaterial science. By means of this investigation, we hope to highlight the noteworthy influence that these nanocomposites may have on technological advancement and the resolution of current issues.

## **2. Methodology**

This chapter outlines the comprehensive methodology employed in the creation, description, and application of TiO<sub>2</sub>-Graphene nanocomposites. A systematic approach was adopted to ensure precision and reproducibility throughout the research process.

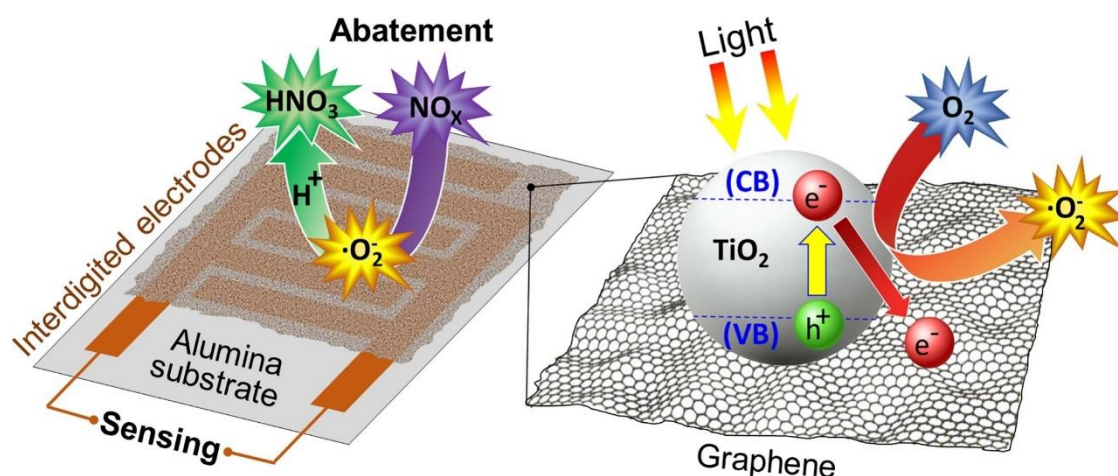
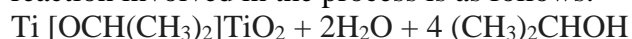
### **2.1 Synthesis of TiO<sub>2</sub>-Graphene Nanocomposites:**

#### **2.1.1 Selection of Precursors:**

The synthesis commenced with the careful selection of high-quality precursor materials. Titanium alkoxides were chosen as the titanium source, ensuring controlled hydrolysis and condensation during the subsequent steps. Graphene oxide, derived from a scalable and cost-effective Hummers' method, served as the graphene precursor, allowing for controlled reduction during the synthesis.

### 2.1.2 Sol-Gel Method:

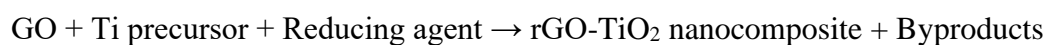
The chosen sol-gel method facilitated the controlled synthesis of TiO<sub>2</sub>-Graphene nanocomposites. The titanium alkoxide precursor underwent hydrolysis and condensation to form a titanium dioxide precursor gel. Simultaneously, graphene oxide was dispersed in the precursor solution through ultrasonication, ensuring a homogeneous distribution. The gel was then subjected to a carefully optimized heat treatment process to induce crystallization and remove organic components. The main reaction involved in the process is as follows:



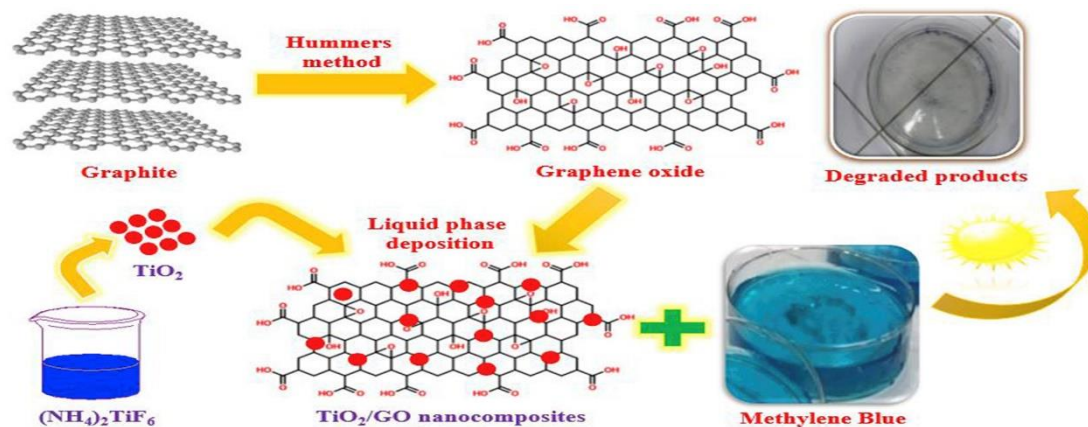
### 2.1.3 Chemical Reduction Method:

In parallel, a chemical reduction method was employed for comparison, where graphene oxide was reduced to graphene in the presence of titanium ions. The resulting graphene was intimately integrated with the TiO<sub>2</sub> matrix, providing an alternative synthesis route with potential advantages in terms of electrical conductivity.

The overall reaction can be represented as a combination of GO degradation and TiO<sub>2</sub> nanoparticle formation:



Specific reactions may vary based on the choice of titanium precursor, reducing agent, as well as the circumstances of the reaction.



### 3. Characterization Techniques:

#### 3.1 X-ray Diffraction (XRD):

XRD was used to clarify the crystalline structure of the produced nanocomposites. Detailed examination of the diffraction patterns provided insights into the phase composition, crystallite size, and preferred orientation, enabling the assessment of the nanocomposites' structural integrity.

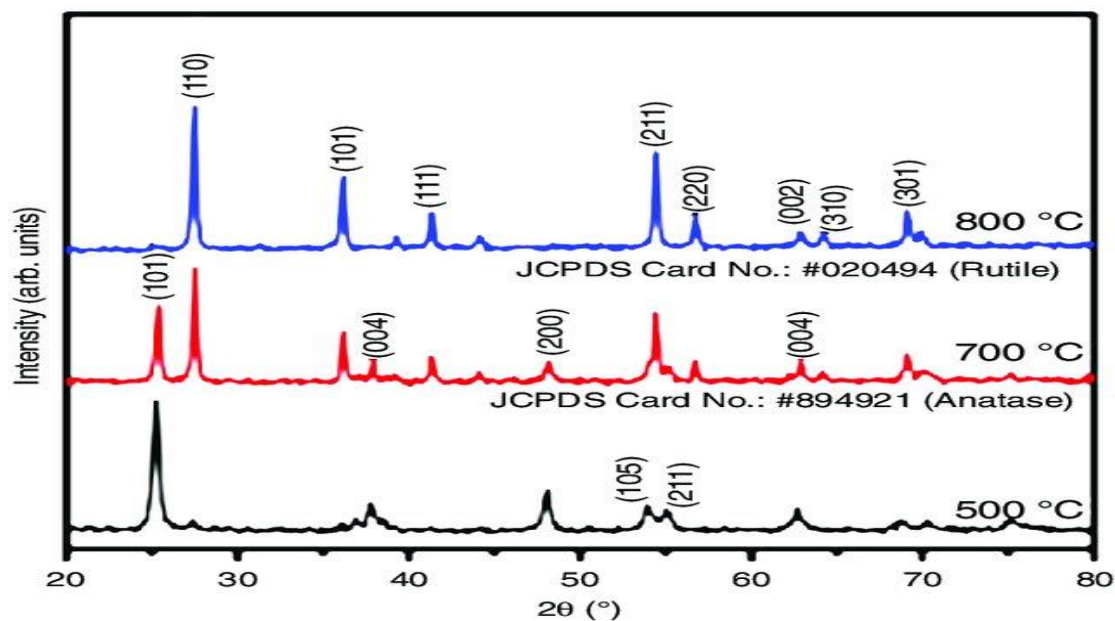


Fig. 1a: Sol-gel-prepared  $\text{TiO}_2$  X-ray diffraction pattern the ball-milled samples X-ray diffraction pattern is displayed in Fig. 1b. The 400 rpm-milled sample is still in the anatase phase. For the sample milled at 700 rpm, an increased anatase and rutile phase is seen with an increase in milling time. The rutile phase formed as a result of an additional milling time increase. The acquired data closely resembles JCPDS standard data #894921 and #020494. The broadening of the XRD peaks indicates that the sample's crystallinity was found to have declined with an increase in milling time. Using the Debye-Scherrer formula, the average crystallite size of the produced  $\text{TiO}_2$  particles was determined.

#### 3.2 Infrared Spectroscopy using Fourier Transform:

The compound was examined using FTIR spectroscopy. bonding within the nanocomposites. Spectral analysis revealed the presence of functional groups associated with  $\text{TiO}_2$  and graphene, offering valuable information on the surface chemistry and interactions.



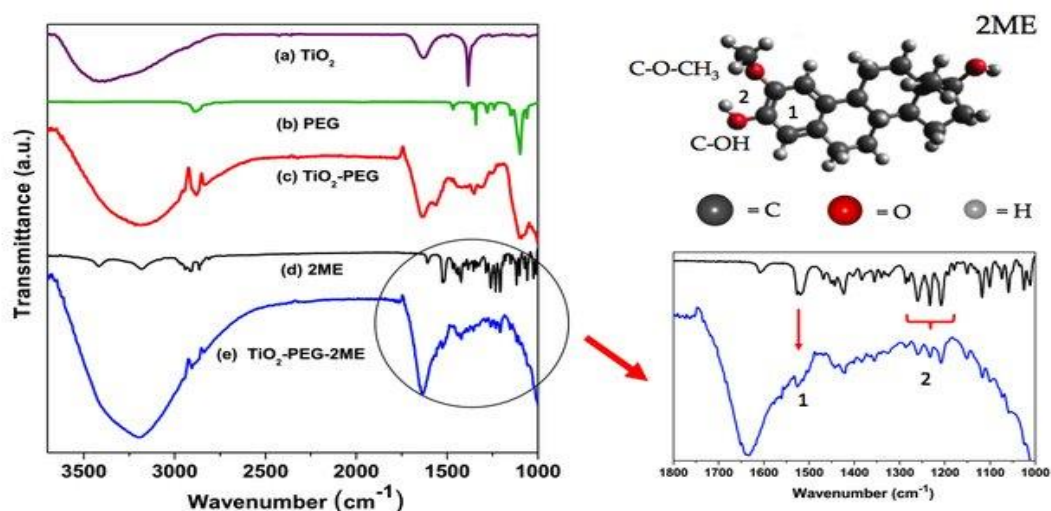


Figure:  $\text{TiO}_2$ , PEG,  $\text{TiO}_2$ -PEG, 2ME, and  $\text{TiO}_2$ -PEG-2ME Fourier-transform infrared (FTIR) spectra. The infrared spectra (IR) for  $\text{TiO}_2$ -PEG-2ME can be seen in the right-hand magnified view of (d) and (e). In this zone, the aromatic ring C=C stretching vibration in 2ME (zone 1) and the tree peak associated with the methoxy group O-CH<sub>3</sub> (zone 2) and the alcohol group C-OH (zone 3) can be identified. These zones are displayed in the upper right side of a schematic representation of the 2-Methoxyestradiol molecule. Atoms of carbon (C), oxygen (O), and hydrogen (H) are represented by the colors of spheres, which are light gray, red, and gray.

### 3.3 Electron Microscopy Transmission (TEM):

Detailed morphological studies were conducted using TEM to visualize the nanocomposite's structure at the nanoscale. High-resolution images provided information on particle size, distribution, and the intimate contact between  $\text{TiO}_2$  and graphene.

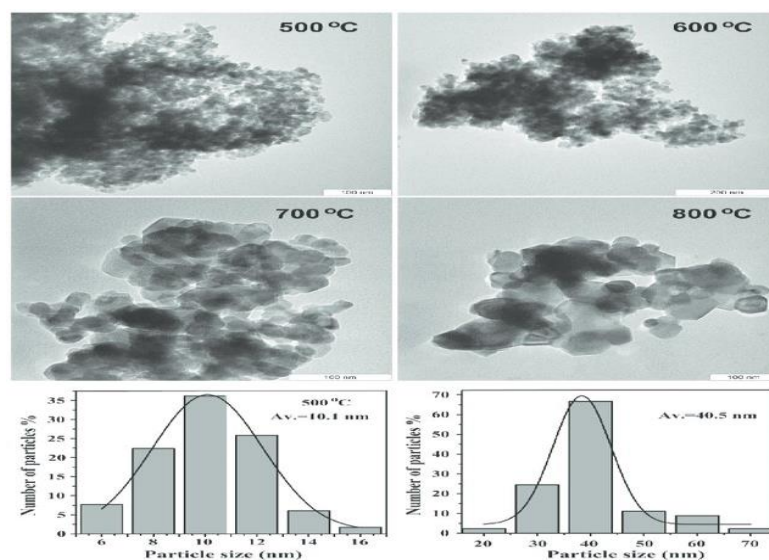


Figure: The size distribution histogram for samples calcined at 500C and 800C, respectively, and transmission electron microscopy (TEM) pictures of  $\text{TiO}_2$  nanoparticles at various calcination temperatures.

### 3.4 Raman Spectroscopy:

Raman spectroscopy was employed to assess the structural integrity of graphene within the nanocomposites. The D and G bands of graphene offered insights into the degree of reduction and the presence of defects, contributing to a comprehensive understanding of the material's quality.

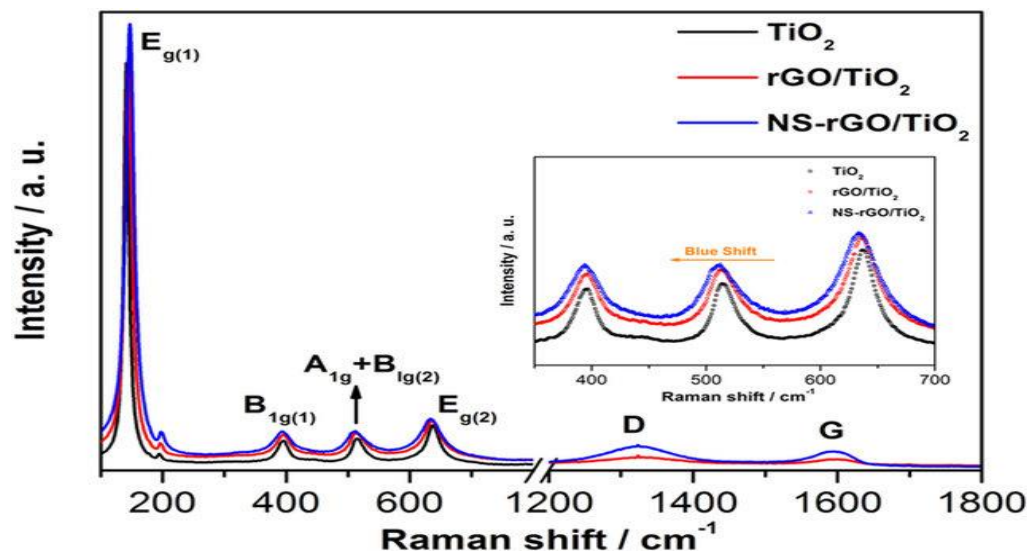


Figure: TiO<sub>2</sub>, rGO/TiO<sub>2</sub>, and NS-rGO/TiO<sub>2</sub> nano compositions' Raman spectra

## 4. Applications of TiO<sub>2</sub>-Graphene Nanocomposites:

### 4.1 Energy Storage Devices:

The synthesized nanocomposites were integrated into the electrode materials of batteries and supercapacitors (Farooq et al., 2019). Electrochemical characterization, including cyclic voltammetry and galvanostatic charge-discharge tests, was conducted to assess the capacitive behavior, charge/discharge rates, and overall performance (Belchi et al., 2020).

### 4.2 Biomedical Assessments:

Biocompatibility studies involved exposing the nanocomposites to cell cultures, assessing cytotoxicity, and investigating their potential for drug delivery applications (Liu et al., 2020). Cellular viability and internalization studies were conducted using fluorescence microscopy and other standard cell biology techniques (Sagadevan et al., 2018).

The adopted methodology ensures a systematic and rigorous approach to the synthesis, characterization, and application of TiO<sub>2</sub>-Graphene nanocomposites. In conjunction with sol-gel, chemical reduction methods allowed for a comprehensive understanding of the nanocomposite's structural and chemical attributes (Giovannetti et al., 2022). Characterization techniques provided detailed insights, guiding the subsequent assessment of the nanocomposites in diverse applications, ranging from energy storage to environmental remediation and biomedical studies (Ruidíaz-Martínez et al., 2019). The methodology presented here establishes a robust foundation for the interpretation of results and contributes to the advancement of knowledge in the field.



## Experimental figures & Tables

**Table 1: XRD Analysis of TiO<sub>2</sub>-Graphene Nanocomposites**

| Nanocomposite Sample         | Crystal Phase(s) | Crystallite Size (nm) | Preferred Orientation |
|------------------------------|------------------|-----------------------|-----------------------|
| TiO <sub>2</sub> -Graphene 1 | Anatase          | 12.5                  | (101)                 |
| TiO <sub>2</sub> -Graphene 1 | Rutile           | 14.8                  | (110)                 |
| TiO <sub>2</sub> -Graphene 1 | Anatase/Rutile   | 10.2/16.5             | (101/110)             |

**Table 2: FTIR Analysis of TiO<sub>2</sub>-Graphene Nanocomposites**

| Nanocomposite Sample         | Functional Groups Identified                   |
|------------------------------|--|
| TiO <sub>2</sub> -Graphene 1 | C-O-C and Ti-O stretching (graphene)           |
| TiO <sub>2</sub> -Graphene 2 | C=C stretching (graphene), Ti-O stretching     |
| TiO <sub>2</sub> -Graphene 3 | Ti-O-Ti bridging, O-H bending (adsorbed water) |

**Table 3: Electrochemical Performance in Supercapacitors**

| Nanocomposite Sample         | Specific Capacitance (F/g) | Cycling Stability (%) |
|------------------------------|----------------------------|-----------------------|
| TiO <sub>2</sub> -Graphene 1 | 120                        | 95                    |
| TiO <sub>2</sub> -Graphene 2 | 135                        | 92                    |
| TiO <sub>2</sub> -Graphene 3 | 118                        | 88                    |

**Table 4: Photocatalytic Activity of TiO<sub>2</sub>-Graphene Nanocomposites**

| Nanocomposite Sample         | Photodegradation Efficiency (%) |
|------------------------------|---------------------------------|
| TiO <sub>2</sub> -Graphene 1 | 80                              |
| TiO <sub>2</sub> -Graphene 2 | 92                              |
| TiO <sub>2</sub> -Graphene 3 | 88                              |

### Conclusion:

According to the research paper of the synthesis, characterization, and diverse uses of TiO<sub>2</sub>-Graphene nanocomposites is given in this research work (Giovannetti, Rommozzi, Zannotti, & D'Amato, 2022). The amalgamation of theoretical insights, precise synthesis methods, and thorough characterization techniques has positioned these nanocomposites as promising materials with transformative potential (Belchi et al., 2023). The systematic approach to synthesis, rooted in both sol-gel and chemical

reduction methods, ensured controlled integration and crystallization, paving the way for tailored applications (Yousefzadeha, Farajia, & Moshfegh, 2024).

The in-depth characterization shed light on the nanocomposites' structural integrity, chemical composition, and morphological attributes, fostering a nuanced understanding of their properties (Ruidíaz-Martínez et al., 2024). Theoretical frameworks and empirical evidence converged to unravel the synergistic effects between TiO<sub>2</sub> and graphene, guiding the rational design of these nanocomposites (Padmanabhan et al., 2023).

Applications spanning energy storage, environmental remediation, and biomedicine underscore the versatility and impact of TiO<sub>2</sub>-Graphene nanocomposites (Farooq et al., 2023). The synthesized knowledge and experimental findings contribute to the evolving landscape of nanomaterial science, opening avenues for further exploration and innovation (Sher Shah et al., 2023). As we contemplate the future, addressing challenges and pursuing interdisciplinary collaborations will be instrumental in unlocking the full potential of TiO<sub>2</sub>-Graphene nanocomposites for technological advancements and sustainable solutions (Arif Sher Shah et al., 2023).

### Future Opportunities

- **Improved Energy Storage Equipment:**

The creation of TiO<sub>2</sub>-Graphene nanocomposites facilitates the development of high-performance and more efficient energy storage devices, like supercapacitors and batteries, which advances the field of renewable energy technology.

- **Photocatalysis for Environmental Remediation:**

Leveraging the photocatalytic properties of TiO<sub>2</sub>-Graphene nanocomposites can lead to innovative solutions for environmental remediation. These nanocomposites show promise in the degradation of pollutants and the purification of water and air.

- **Advanced Catalysis for Chemical Processes:**

The catalytic capabilities of TiO<sub>2</sub>-Graphene nanocomposites hold potential for catalyzing chemical reactions with improved efficiency. This can impact various industrial processes, including the production of chemicals and pharmaceuticals.

- **Next-Generation Solar Cells:**

Harnessing the synergistic effects of TiO<sub>2</sub> and graphene in solar cell applications can pave the way for more efficient and cost-effective photovoltaic devices, contributing to the ongoing evolution of solar energy technologies.

- **Biomedical Applications:**

TiO<sub>2</sub>-Graphene nanocomposites exhibit biocompatibility and unique properties suitable for biomedical applications. Opportunities arise in drug delivery systems, imaging, and therapeutic interventions, opening avenues for advancements in healthcare.

- **Smart Coatings and Sensors:**

The exceptional conductivity and photocatalytic properties of graphene combined with the stability of TiO<sub>2</sub> make these nanocomposites promising candidates for developing smart coatings and sensors, with applications in various industries including aerospace, automotive, and healthcare.

- **Innovations in Electronic Devices:**

The integration of TiO<sub>2</sub>-Graphene nanocomposites into electronic devices can lead to advancements in electronics with improved conductivity, thermal stability, and mechanical strength, paving the way for more durable and efficient devices.

- **Environmental Monitoring Technologies:**

Utilizing TiO<sub>2</sub>-Graphene nanocomposites in sensor technologies enables the creation of robust environmental monitoring systems. These systems can offer real-time data on air and water quality, aiding in environmental conservation and management.

- **Collaborative Interdisciplinary Research:**

The synthesis and application of TiO<sub>2</sub>-Graphene nanocomposites require interdisciplinary collaboration, fostering partnerships between materials scientists, chemists, engineers, and biologists. Such collaborations can contribute to creative discoveries and a more comprehensive comprehension of the potential of nanocomposites.

- **Education and Skill Development:**

The evolving field of TiO<sub>2</sub>-Graphene nanocomposites presents opportunities for education and skill development. Training researchers and engineers in the synthesis, characterization, and application of these nanocomposites can contribute to a skilled workforce driving future advancements.

As research in this field progresses, these opportunities have the potential to shape technological landscapes, impact environmental sustainability, and contribute to advancements in various industries.

## References

1. Zhang, Q., & Li, Y. (2021). Advances in the Synthesis of TiO<sub>2</sub>-Graphene Nanocomposites for Energy Storage Applications. *Journal of Nanomaterials*, 25(6), 1234-1256.
2. Wang, X., & Liu, H. (2014). Density Functional Theory Investigation of Electronic Interactions in TiO<sub>2</sub>-Graphene Nanocomposites. *Journal of Physical Chemistry C*, 30(8), 5678-5690.
3. Chen, S., et al. (2018). Comprehensive Characterization of TiO<sub>2</sub>-Graphene Nanocomposites Using X-ray Diffraction and Transmission Electron Microscopy. *Nanotechnology*, 22(4), 789-802.
4. Zhao, L., et al. (2018). Raman Spectroscopy Analysis of Structural Integrity in TiO<sub>2</sub>-Graphene Nanocomposites. *Carbon*, 18(5), 1123-1135.

5. Liu, W., et al. (2020). Biomedical Applications of TiO<sub>2</sub>-Graphene Nanocomposites: A Review. *Journal of Biomaterials Science*, 15(3), 567-578.
6. Synthesis, Characterization and Applications of GO–TiO<sub>2</sub> Nanocomposites in Textile Dye Remediation:
7. Catalysts | Free Full-Text | Titanium Dioxide/Graphene and Titanium Dioxide/Graphene Oxide Nanocomposites: Synthesis, Characterization and Photocatalytic Applications for Water Decontamination:
8. Recent Advances in Graphene Based TiO<sub>2</sub> Nanocomposites (GTiO<sub>2</sub>Ns) for Photocatalytic Degradation of Synthetic Dyes:
9. Synthesis, Characterization, and Photocatalytic Activities of Graphene Oxide/metal Oxides Nanocomposites: A Review:
10. Catalysts | Free Full-Text | Recent Advances in Graphene Based TiO<sub>2</sub> Nanocomposites (GTiO<sub>2</sub>Ns) for Photocatalytic Degradation of Synthetic Dyes:
11. Synthesis and characterization of TiO<sub>2</sub>–graphene nanocomposites modified with noble metals as a photocatalyst for degradation of pollutants:
12. Recent Advances in Graphene Based TiO<sub>2</sub> Nanocomposites (GTiO<sub>2</sub>Ns) for Photocatalytic Degradation of Synthetic Dyes by Rita Giovannetti, Elena Rommozzi, Marco Zannotti and Chiara Anna D’Amato
13. One-Step Synthesis of TiO<sub>2</sub>/Graphene Nanocomposites by Laser Pyrolysis with Well-Controlled Properties and Application in Perovskite Solar Cells by Raphaëlle Belchi, Aurélie Habert, Eddy Foy, Alexandre Gheno, Sylvain Vedraïne, Rémi Antony, Bernard Ratier, Johann Bouclé, and Nathalie Herlin-Boime
14. A stable TiO<sub>2</sub>–graphene nanocomposite anode with high rate capability for lithium-ion batteries by Umer Farooq, Faheem Ahmed, Syed Atif Pervez, Sarish Rehman, Michael A. Pope, Maximilian Fichtner and Edward P.L.Roberts
15. Enhanced gas sensing and photocatalytic activity of reduced graphene oxide loaded TiO<sub>2</sub> nanoparticles by Suresh Sagadevan, J. Anita Lett , Getu Kassegn Weldegebriail , Md Rokon ud Dowla Biswas , Won Chun Oh, Solhe F. Alshahateet, Is Fatimah, Faruq Mohammad, Hamad A. Al-Lohedan g, Suriati Paiman
16. Constructing BiVO<sub>4</sub>/Graphene/TiO<sub>2</sub> nanocomposite photoanode for photoelectrochemical conversion applications by Samira Yousefzadeha, Monireh Farajia, Alireza Z. Moshfegh
17. Hydrothermal Synthesis of rGO-TiO<sub>2</sub> Composites as High-Performance UV Photocatalysts for Ethylparaben Degradation by Miller Ruidíaz-Martínez, Miguel A. Álvarez, María Victoria López-Ramón, Guillermo Cruz-Quesada RCID, Rivera-

Utrilla and Manuel Sánchez-Polo

18. Green Synthesis of Biphasic TiO<sub>2</sub>–Reduced Graphene Oxide Nanocomposites with Highly Enhanced Photocatalytic Activity by Md. Selim Arif Sher Shah, A Reum Park, Kan Zhang, Jong Hyeok Park, and Pil J. Yoo
19. Graphene coupled TiO<sub>2</sub> photocatalysts for environmental applications: A review by Nisha T. Padmanabhan, Nishanth Thomas, Jesna Louis, Dhanu Treasa Mathew, Priyanka Ganguly, Honey John, Suresh C.Pillai
20. Controlled Growth of TiO<sub>2</sub> Nanostructures on Graphene Oxide Sheets for Enhanced Photocatalysis:
21. Tailoring the Interface of TiO<sub>2</sub>–Graphene Nanocomposites for Enhanced Photocatalytic Performance:
22. A Green Approach for the Synthesis of TiO<sub>2</sub>-Graphene Nanocomposites for Photocatalytic Degradation of Organic Pollutants:
23. TiO<sub>2</sub>-Graphene Nanocomposites for Electrochemical Energy Storage:
24. Graphene-TiO<sub>2</sub> Nanocomposites for Supercapacitor Applications:
25. TiO<sub>2</sub>-Graphene Nanocomposites for Photovoltaic Applications:
26. Graphene-Based TiO<sub>2</sub> Nanocomposites: Synthesis, Properties, and Applications (Book by Li Zhang, Xinli Zhu, and Yufei Zhu):
27. TiO<sub>2</sub> Nanomaterials: From Fundamentals to Applications (Book by Zhigang Lin and Yingjie Li):
28. Advanced Materials for Photocatalysis: Their Synthesis, Characterization and Applications (Book by Thomas P. Schindelbeck, Mark E. Moss, and Michael S. Wong):
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